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Key indicators

Single-crystal X-ray study
 $T = 193$ K
Mean $\sigma(\text{Si}-\text{C}) = 0.007$ Å
 R factor = 0.061
 wR factor = 0.104
Data-to-parameter ratio = 12.5For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Tris[bis(trimethylsilyl)amido]samarium
tetrahydrofuran solvate

The molecular structure of the title complex, $[\text{Sm}\{\text{N}(\text{SiMe}_3)_2\}_3] \cdot \text{C}_4\text{H}_8\text{O}$, an important key compound in lanthanide chemistry, has been determined. The crystal structure analysis reveals a trigonal–pyramidal structure, isotype to other structurally characterized lanthanide tris-silylamides. The molecule is disordered due to the D_3 site symmetry.

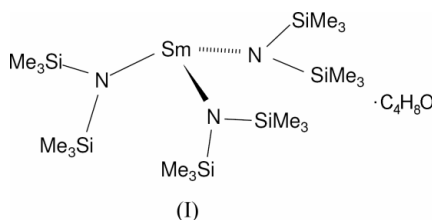
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Comment

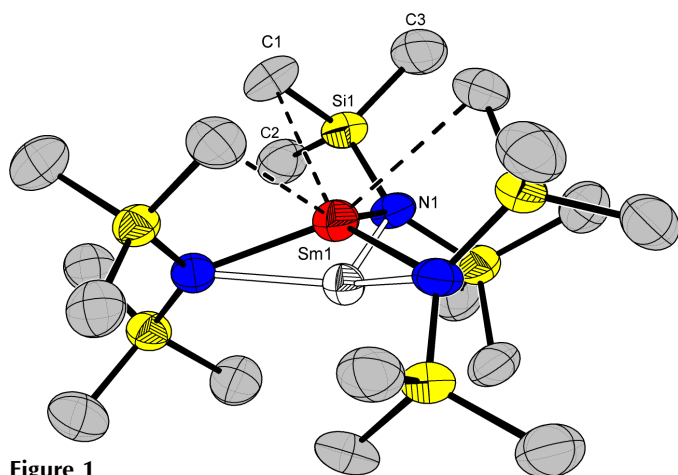
Homoleptic silylamides of the rare earth elements are well established starting materials for the synthesis of more complex functionalized compounds of the lanthanides *via* amine metathesis reactions with OH-, NH- and CH-acidic ligands (Anwander, 1996). The title compound, (I), was first described by Bradley *et al.* (1973); however, its molecular structure has never been established. It serves as a catalyst for the ring-opening polymerization of lactones (Agarwal *et al.*, 1999). Samarium, with an Sm^{3+} radius of 1.13 Å, is the largest of the lanthanide(III) cations and we were therefore interested to compare the molecular structure with those of other $[\text{Ln}\{\text{N}(\text{SiMe}_3)_2\}_3]$ compounds (Niemeyer, 2002, and references therein; Hitchcock, 2004).



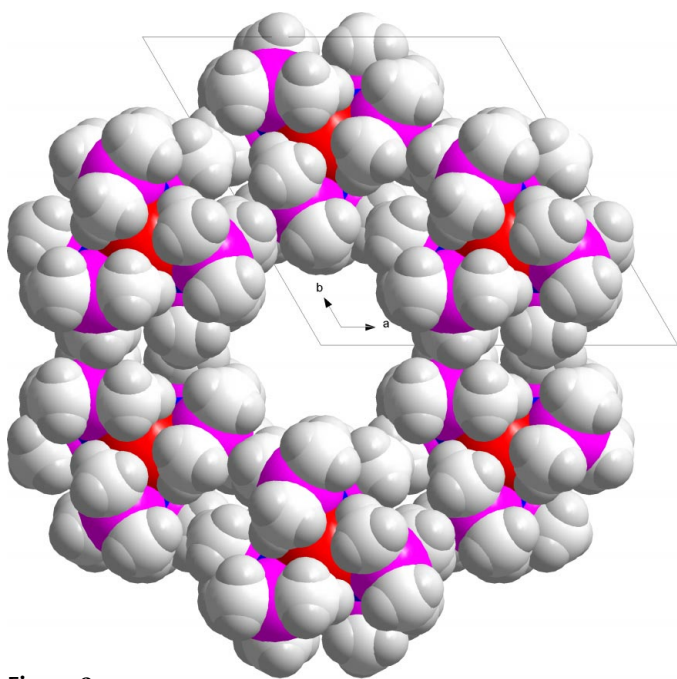
Compound (I) crystallizes in the trigonal space group $P\bar{3}1c$ with $Z = 2$ and is isotypic to the other $[\text{Ln}\{\text{N}(\text{SiMe}_3)_2\}_3]$ compounds and therefore shows similar structural features: (i) the three-coordinate pyramidal Sm atom is disordered and adopts a position on a threefold axis 0.515 (2) Å above and below the N_3 plane; (ii) there are short ‘agostic’ interactions between Sm and the trimethylsilyl groups, with $\text{Sm1} \cdots \text{C1} = 2.988$ (6) Å [see Niemeyer (2002) for details about this interaction]; (iii) heavily disordered solvent (tetrahydrofuran) is present in solvent-accessible channels.

Experimental

The title compound was prepared using $\text{NaN}(\text{SiMe}_3)_2$ instead of $\text{LiN}(\text{SiMe}_3)_2$ in a modification of a literature procedure (Bradley *et al.*, 1973). SmCl_3 (3.76 g, 14.63 mmol) was suspended at 273 K in THF (70 ml) and a solution of $\text{NaN}(\text{SiMe}_3)_2$ (8.05 g, 43.86 mmol) in THF (30 ml) was added over a period of 2–3 min. After stirring for 24 h at room temperature, all solvent was removed under reduced pressure. The residue was extracted from a glass frit with 5×50 ml boiling


Figure 1

A view of (I) showing the Sm...C contacts (dashed lines). Displacement ellipsoids are drawn at the 50% probability level. H atoms and the disordered tetrahydrofuran molecule have been omitted. Atoms of the asymmetric unit are labelled.


Figure 2

Space-filling packing diagram of (I), showing a solvent-accessible channel.

hexane. The extract was evaporated to dryness and the residue was sublimed at 363 K/10⁻⁴ mbar (yield: 6.85 g, 74%, literature 50–70%; yellow, crystalline; m.p. 431–432 K, literature m.p. 428–431 K). ¹H NMR (200.1 MHz, C₆D₆): δ -1.58. ¹³C NMR (50.3 MHz, C₆D₆): δ = 2.2. Single crystals were grown by cooling a hexane/THF solution.

Crystal data

[Sm(C₆H₁₈NSi₂)₃]·C₄H₈O
M_r = 703.64
 Trigonal, *P*3̄1*c*
a = 16.490 (3) Å
c = 8.3485 (11) Å
V = 1966.0 (6) Å³
Z = 2
D_x = 1.189 Mg m⁻³

Mo *K*α radiation
 Cell parameters from 2000 reflections
 $\theta = 2-25^\circ$
 $\mu = 1.69 \text{ mm}^{-1}$
T = 193 (2) K
 Needle, colourless
 0.33 × 0.06 × 0.06 mm

Data collection

Stoe IPDS-2 diffractometer
 ω scans
 Absorption correction: multi-scan
 (Blessing, 1995)
*T*_{min} = 0.855, *T*_{max} = 0.903
 8924 measured reflections
 1154 independent reflections

784 reflections with *I* > 2σ(*I*)
*R*_{int} = 0.118
 $\theta_{\text{max}} = 25.0^\circ$
h = -19 → 19
k = -19 → 13
l = -8 → 9

Refinement

Refinement on *F*²
R[*F*² > 2σ(*F*²)] = 0.061
wR(*F*²) = 0.104
S = 1.04
 1154 reflections
 92 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0393P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.34 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.92 \text{ e \AA}^{-3}$
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.0145 (16)

Table 1

Selected geometric parameters (Å, °) for (I).

Sm1–N1	2.294 (6)	Si1–C3	1.861 (7)
Sm1–C1	2.988 (6)	Si1–C1	1.873 (6)
Si1–N1	1.709 (3)	Si1–C2	1.875 (6)
N1 ⁱⁱ –Sm1–N1	115.11 (3)	C3–Si1–C2	107.6 (3)
N1–Si1–C3	113.5 (3)	C1–Si1–C2	107.1 (3)
N1–Si1–C1	108.2 (3)	Si1–N1–Si1 ⁱⁱⁱ	126.1 (4)
C3–Si1–C1	106.7 (3)	Si1–C1–Sm1	79.8 (2)
N1–Si1–C2	113.3 (2)		

Symmetry codes: (i) 1 - *y*, 1 - *x*, -½ - *z*; (ii) 1 - *y*, *x* - *y*, *z*; (iii) *x*, *x* - *y*, -½ - *z*.

There was substantial residual electron density present in the solvent-accessible channels in the structure of the title compound. A complete tetrahydrofuran molecule was modelled with restraints for 1,2- and 1,3-distances and was anisotropically refined with restraints for the displacement parameters. A common occupation factor refined to 0.171 (5) and was fixed at 0.16667. This leads to two solvent molecules in the unit cell with a sixfold disorder due to crystallographic symmetry. This heavy disorder and the disorder of the Sm atom may explain the weakness of the data set (*R*_σ = 0.081) and the relatively high *R*_{int} value (0.118). All H atoms were included in calculated positions and refined using a riding model [C–H = 0.96 Å and *U*_{iso}(H) = 1.5*U*_{eq}(C) for CH₃ atoms, and C–H = 0.99 Å and *U*_{iso}(H) = 1.2*U*_{eq}(C) for CH₂ atoms]. Although a refinement in the non-centrosymmetric space group *P*3̄1*c* gave slightly better *R* values, it was finally not successful. The same disorder for the Sm atom and for the tetrahydrofuran molecule was present as in space group *P*3̄1*c*. There were large correlations between parameters, high standard deviations and a distorted geometry. Some displacement parameters of the tetrahydrofuran molecule were non-positive definite.

Data collection: *X-AREA* (Stoe & Cie, 2003); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 2001); software used to prepare material for publication: *WinGX* publication routines (Farrugia, 1999).

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